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EFFECT OF GINGER EXTRACT ON THE PHYSICOCHEMICAL AND SENSORY PROPERTIES OF YOGHURT

^{1,*}Njoya Moyouwou Amadou, ²Ejoh Aba Richard, ³Kuiate Jules – Roger, ¹Nain Caroline Waingeh, ²Kuboh Delphine Ateh, ²Audra Faison Mbiydzengeh, ¹Nde Sylvanus Che and ¹Mahbou Peter Yunenyui

¹Food Technology and Post–Harvest laboratory, Irad–Bambui; P.O.Box 51 or 80 Bamenda, Cameroon ²Department of Food and Bioressource Science and Technology, Coltech – University of Bamenda; P.O. Box 39 Bambili, Cameroon

³Department of Biochemistry, Faculty of Science – University of Dschang; P.O. Box 67 Dschang, Cameroon

ARTICLE INFO ABSTRACT Article History: The aim of this study was to evaluate the effect of ginger extracts on the physicochemical and sensory properties of

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Yoghurt, ginger extracts, Physicochemical And sensory Properties, storage.

The aim of this study was to evaluate the effect of ginger extracts on the physicochemical and sensory properties of yoghurt. Six ginger extracts were obtained by decoction, maceration at room temperature and at refrigerated conditions ($4 - 6^{\circ}$ C) of ginger paste and powder. Yoghurt were obtained by adding each extract at0%, 5%, 10% and 15% (V/V). Physicochemical properties of yoghurt samples (pH, titratable acidity, dry matter, ash, fat, and non-fat solid) were determined at day 0. The pH and titratable acidity were also evaluated during 30 days of storage at refrigerated conditions at 5 days' intervals. The sensory attributes assessed were colour, odour, taste, texture and overall acceptability. From the results,ginger extractsdid not affect the pH of yoghurt but decreased the titratable acidity, dry matter, fat, non-fat solid (NFS) and ash at high concentrations. Ginger extracts did not also affect the reduction of the level of appreciation of yoghurt by the panellist but at 5%, no effect was observed. Spicing the yoghurt with 5% of ginger extract during its manufacturing could therefore be recommended for human consumption.

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INTRODUCTION

Yoghurt is the most popular fermented dairy product all over the word (Tamine and Deeth, 1980; Madhu *et al.*, 2012). It is a semi - solid product obtained through lactic acid fermentation of milk. During fermentation, lactose of milk is transformed into lactic acid by lactic acid bacteria known as *Lactobacillus bulgaricus* and *Streptococcus thermophilus*. Yoghurt represents an excellent source of proteins, minerals and vitamins (Buttris, 1997). It is more digestible than milk due to the partial or total hydrolysis of fat, lactose and proteins and therefore recommended for people with lactose intolerance or milk proteins allergy. Moreover, yoghurt endowed health benefits including reduction of risk of diseases and improvement of health gut, lowering blood cholesterol level, improvement of immune response and helping assimilation of protein, calcium and iron (Perdigeon et al., 1998; Heymans, 2000; Marona and Pedrigon, 2004). Yoghurt quality depends on the change observed on its physiochemical, microbiological and sensory characteristics. In addition, its shelf-life is limited due to the deterioration of its properties during storage. In order to improve the yoghurt quality, different flavouring ingredients have been added during its manufacturing. Mostly, artificial flavours are used while, the use of natural flavours could be more benefits to the consumer. Addition of herbs could be an effective strategy to improve functionality of milk and milk products with respect to the health benefits, food safety andbio-preservation (Aswal et al., 2012). Thus, natural fruits as flavour have contributed to improve the level of consumption of yoghurt due to their nutritive value in addition to their health benefits, flavour and aroma. The use of spices as flavour during yoghurt manufacturing could also be useful considering the fact that, they are plants rich in bioactive components with health benefits and constitute a potential source of minerals and vitamins. In addition, they are

^{*}Corresponding author: Njoya Moyouwou Amadou,

Food Technology and Post–Harvest laboratory, Irad–Bambui; P.O. Box 51 or 80 Bamenda, Cameroon.

characterised by their antimicrobial properties and could consequently positively affect the shelf life of yoghurt. Ginger (Zingiberofficinale, Roscoe) is a spice commonly used culinary for its taste, aroma and flavour. Traditionally, it aids against gastrointestinal disorders as antispasmodic in stomach aches and diarrhoea and has been used since ancient times as traditional medicines in Asia. It is characterised by its pungent taste and represents a potential source of fibres, proteins, minerals and essential amino acids (Latonaet al., 2012; Olubunmi et al., 2013). It has many bioactive components, and amongst gingerols, the main active component, the 6-gingerol is the most important (ICMR, 2003). It helps in the treatment and prevention of many types of cancer. Many studies indicated that ginger is endowed with antibacterial and antifungal properties (Gugnani and Ezenwanze, 1985; Akoachere et al., 2002; Atai et al., 2009; Kaushik and Goyal, 2011); hypoglycaemia, hypolipidemia and hypocholesterolaemia properties (Ahmed and Sharma, 1997; bhandari et al., 2005; Al-Amin et al., 2006; Mendi, 2011) and antioxidant and anti-inflammatory effects (Aruomaet al., 1997; Thomson et al., 2002). It also improves gastro-intestinal function (Guyer, 2003; ICMR, 2003). Furthermore, ginger and its extracts have pharmacological, digestion stimulant and antimicrobial activities. Due to its antimicrobial nature, ginger could increase the shelf-life of foods (Adesokan et al., 2010). The use of ginger to spice yoghurt could therefore be benefit and render the yoghurt more functional. Thus, the present study is to investigate on the effect of ginger extracts on the physicochemical and sensory properties of yoghurt.

MATERIAL AND METHODS

Preparation of ginger powder: The roots (rhizomes) of ginger (*Zingiberofficinale*) from Bafut (Bafut Sub-Division, Mezam Division, North-West Region, Cameroon) were purchased from the Bamenda Food Market in the North-West Region of Cameroon and brought to the Food Technology Laboratory of the Regional Centre of IRAD - Bambui, Cameroon. The ginger was washed several times with tap water (Potable water). Then, it was peeled, rewashed using tap water, sliced into small sizes of 2- 3mm diameter thick and dried in a vacuum oven at $60 - 65^{\circ}$ C for 72hrs (Olubunmi *et al.*, 2013). The dried ginger was blended into powder using a sifter of pore size $\leq 300 \mu$ m. The sieved ginger powder was put in plastic bags, stored at room temperature under unlight.

Preparation of ginger paste: The ginger paste was obtained after successively washing, peeling, rewashing and grating of the ginger roots.

Preparation of ginger aqueous extract: Ginger extracts was obtained from ginger powder and ginger paste by modified methods described by Kaushik and Goyal (2011) and Abd El – Aziz *et al.* (2015).

Decoction: Decoction was done by boiling 100g of ginger paste in 3.21 of potable water (drinkable water)till one fourth (1/4) of the initial volume was attained. The solution was filtered twice using a muslin cloth, then allowed at room temperature for sedimentation and the supernatant collected. The same procedure was used with ginger powder but using 100g in 41 of drinkable water.

Maceration: Maceration was done by soaking 100g of ginger paste for 1h in 800ml of hot drinkable water firstly boiled and cooled at 80°C for 1hr. After cooling at room temperature, the mixture was then allowed at refrigerated conditions (4 - 6°C) for three (03) days and filtered twice using a muslin cloth. Maceration was also done at room temperature for three (03) days. Ginger powder was also macerated as indicated above with ginger paste but using 100g of powder in 1l of drinkable water. All the extracts collected was pasteurised at 75°C/3 – 5s, cooled at room temperature and kept in the freezer at \leq 0°C for further uses.

The following extracts were obtained

- Extract 1: Maceration of ginger paste at refrigerated conditions
- Extract 2: Decoction of ginger paste
- Extract 3: Maceration of ginger paste at room temperature
- Extract 4: Decoction of ginger powder
- Extract 5: Maceration of ginger powder at refrigerated conditions
- Extract 6: Maceration of ginger powder at room temperature.

Preparation of ginger spiced yoghurt: Fresh cow's milk (Physicochemical composition given in table 1) collected from the dairy unit of the research centre was used to produce yoghurt samples in five replicates. The yoghurt was made according to the modified method of Lee and Lucey (2010). The milk was pasteurized by heating at 85-90°C for five minutes in a boiling water bath during which 6.5% (w/v) of sugar was added. The milk was then rapidly cooled to inoculation temperature $(42 - 45^{\circ}C)$ followed by addition of 2.5% (w/v) yoghurt starter culture (CHR HANSEN YF -L811) comprising Streptococcus *thermophilus* and Lactobacillus bulgaricus in a 1:1 ratio and incubation at 42 -45°C for 3h when the yoghurt was set. The set yoghurt was directly cooled in refrigerator $(4 - 6^{\circ}C \text{ for} 06 - 12h)$ before manual stirring. After stirring, each ginger extract has been added to the yoghurt at different concentrations (0,5, 10 and 15% (V/V)). 19 yoghurt samples were then obtained as indicated in Table 2. All the yoghurt samples were kept in the refrigerator (4 - 6°C) for physicochemical analysis and sensory evaluation.

Physicochemical analysis: The physicochemical analysis of each sample was done in duplicate. The dry matter (DM), pH, Titratable Acidity (TA), ash and fat were determined according to the standard Association of Official Analytical Chemists methods (AOAC, 1990). The pH and TA determination were also done within 30 days of storage at refrigerated conditions $(4 - 6^{\circ}C)$ and at 5 days' intervals. The Non-Fat Solid (NFS) was obtained using the formula: NFS (%) = dry matter (%) – fat content (%).

Sensory evaluation: Sensory evaluation of all yoghurt samples was carried out using a five-point hedonic scale with the following as categories: Excellent=5; Very Good=4; Good=3; Fair=2 and Poor=1. Colour (appearance), odour, texture, taste and overall acceptability of yoghurt samples were evaluated by an untrained panel of 30 persons between 20 and 60 years old who were regular yoghurt consumers and made of researchers, techniciansand students on internship of the centre of IRAD – Bambui.

Statistical analysis: Data obtained were expressed as Mean±SD and subjected to the Analysis of variance (ANOVA) using the Statgraphics Plus, version 5.0 statistical package. The means obtained were separated using the Fischer Test (P \leq 0.05).

RESULTS AND DISCUSSION

Physicochemical analysis: The physicochemical composition of the yoghurt samples is shown in Table 3. The results indicate that pH of yoghurt was not affected (P>0.05) by the ginger extracts. This could be due to the aqueous character of extracts. This result is not consistent with those of Njoya et al. (2016); Rashid and Takur (2012) and Manjula et al. (2011) who obtained a reduction of yoghurt pH by addition of pineapple, honey and sapota respectively and this may be due to the fermentable sugars and organic acids present in these products. Moreover, results of Yousef et al. (2013) and Ayar and Gürlin (2014) indicated that the yoghurt pH is also influenced by the pH (degree of acidity) of the ingredient or additive used. The pH values obtained were similar to the normal pH value of yoghurt which is between 4.0 and 4.1 and in accordance with the standardof ≤4.6 (FDA, 2009). They are similar to results obtained by Güler and Park (2011) and Ponka et al. (2013) that did their studies on popular Turkish yoghurt and artisanal Cameroonian yoghurts respectively. However, they are higher than the values proposed by Njoya *et al.* (2016) and Joseph and Joy (2011) that worked on pineapple flavoured voghurt and voghurt present on the Nigerian market respectively and, lower than the values of Isam et al. (2011) and Yousef et al. (2013) who respectively studied the yoghurt from fresh milk (Cow and camel) and fruit flavoured yoghurt. Generally, the pH of the yoghurt depends on many factors including the milk composition, the ingredients used and the activity of lactic acid bacteria. Titratable acidity of yoghurt samples seemed to reduce with increase in ginger extract concentration.

All yoghurt samples spiced with 15% of ginger extract presented a lower (P<0.05) titratable acidity compare to the unspiced yoghurt (control sample) while spiced yoghurt containing 5% of ginger extract except sample O (yoghurt sample containing with 5% of extract 5) were similar (P>0.05) to the control yoghurt sample. The reduction of titratable acidity of yoghurt with the increase of ginger extract concentration could be due to the dilution effect. This tendency was contrary to those observed by Njoya et al. (2016), Yousef et al. (2013), Rashid and Takur (2012) and Manjula et al. (2011). These authors rather obtained increasing of titratable acidity with addition of fruits or honey to yoghurt. Fruits or honey are fermentable sugars sources which serve as substrate to lactic acid bacteria with production of lactic acid in addition to their organic acids components leading to an increase of titratable acidity. The titratable acidity values observed were in accordance with the normal value (0.9 - 1%) of yoghurt and the values proposed by USDA (2001), FDA (2009) and Turkish legislation (Tarakçi and Küçüköner, 2003). They could be higher than results proposed by Njoya et al. (2016) and Joseph and Joy (2011), lower than those of Güler and Park (2011) and similar to those of Elfaki and AbbElrazig (2010) who studied on voghurt samples from Sudan and Yousef et al. (2013). The titratable acidity of yoghurt depends mostly on the activity of lactic acid bacteria which fermented sugars into lactic acid. Thus, it depends of the availability of nutrients (especially fermentable sugars) and water activity.

The unspiced yoghurt showed the highest (P<0.05) dry matter content amongst all the yoghurt samples studied. The increase of ginger extract led to a reduction of yoghurt dry matter and this could be related to the high water content of ginger extract and the dilution effect. This result is in agreement with that of Salwa et al. (2004) who carried out their studies on yoghurt produced by addition of carrot juice. Nevertheless, it was not consistent with those of Niova et al. (2016), Joseph and Joy (2011), Rashid and Takur (2012) and Yousef et al. (2013). Those authors rather used additives (Fruits and honey) in semi - solid form with relative high dry matter which lead to the increase of dry matter content of the yoghurt. Dry matter values obtained were close to those of Njoya et al. (2016), Ponka et al. (2013) and Joseph et al. (2011); lower than values proposed by Rashid and Takur (2012) and higher than results of Elfaki and AbbElrazig (2010), Gûler and Park (2011) and Isam et al. (2011). Generally, the yoghurt dry matter content is due to many factors including the milk composition as well as the quantity and the nature of ingredients used.

The unspiced yoghurt sample presented the highest value of ash content and, these value was comparable (P>0.05) to those of yoghurt spiced with 5% of extracts 4 and 6 (Samples L and R). Moreover, yoghurt ash content was reducing generally with increasing of ginger extract and it could be due to the low ash content of the ginger extract and dilution factor. Previous studies revealed an increase of yoghurt ash content with additives containing relatively high ash content and in regard of the quantity added (Yousef et al., 2013). The values obtained were similar to those of Elfaki and AbbElrazig (2010), Yousef et al. (2013), Ponka et al. (2013) and Njoya et al. (2016) while findings of Güler and Park (2011) indicated lower values. The yoghurt ash content depends on many factors including the milk composition, the quantity and the nature of the ingredients used. The unspiced yoghurt sample showed the highest (P<0.05) fat content. Moreover, the fat content was reducing with increasing of ginger extract concentration in yoghurt. This reduction in fat content could be explained by the low or neglected level of fat content of the aqueous extract of ginger and it is in accordance with previous findings (Njoya et al., 2016; De Silva and Rathnayaka, 2014; Yousef et al., 2013; Rashid and Takur, 2012; Salwa et al., 2004). From those findings, use of additives (fruits and honey) with low fat content during yoghurt manufacturing resulted to a reduction of yoghurt fat content. The yoghurt fat content observed were corroborated with results of Güler and Park (2011) and seemed to be higher than values proposed by Salwa et al. (2004), Yousef et al. (2013), Ponka et al. (2013) and Njoya et al. (2016). The fat content of yoghurt generally depends of the milk composition (Fat content) on one hand and, on the other hand to the composition (fat content) and proportion of additives used.

The Non –Fat Solid (NFS) of the unspicedyoghurt was the highest (P \leq 0.05) amongst all the yoghurt samples. Increasing the ginger extract concentration led to reduce the NFS content of yoghurt and could be related to the low or negligible level of SNF of the extracts and the dilution effect. Manjula *et al.* (2012) and Njoya *et al.* (2016) observed similar result by using fruits with low fat content and consequently high NFS content, though De Silva and Rathnayaka (2014) on its one part, did not observed significant differences in NFS content between fruit spiced and plain yoghurt. The values obtained are \geq 8.5% as recommended by American standards (FDA, 2009). They are similar to results of previous studies (Joseph and Joy, 2011;

Njoya *et al.*, 2016), lower than results obtained by De Silva and Rathnayaka (2014) and higher than values of Manjula *et al.* (2012). The NFS of yoghurt is related to milk composition and the fat content of the ingredient and the quantity used. The differences on physicochemical composition observed

between the results of the present studies and previous ones could be due to the fact that physicochemical properties of yoghurt depend on many factors including milk composition, technology, nature and composition of additives; strains and activity of lactic acid bacteria.

	Table 1. P	Physicochemical	composition	of fresh	cow milk
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				(%)			
pН	ТА	DM	Ash	Fat	Proteins	Lactose	NFS
6.73±0.05	0.45 ± 0.04	13.7±0.23	0.68 ± 0.06	5.1±0.12	3.4±0.11	4.61±0.25	8.7±0.14

TA: Titratable Acidity (%lactic acid); DM: Dry Matter; NFS: Non - Fat Solid

Table 2. Yoghurt samples

Ginger extract	Extract concentration (%)	Yoghurt sample	Ginger extract	Extract concentration (%)	Yoghurt sample
/	0	unspiced	Extract 4	15	J
Extract 1	15	A		10	K
	10	В		5	L
	5	С	Extract 5	15	М
Extract 2	15	D		10	Ν
	10	Е		5	0
	5	F	Extract 6	15	Р
Extract 3	15	G		10	Q
	10	Н		5	R
	5	Ι	/	/	/

Table 3. Physicochemica	l composition	of yoghurt	samples
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Yoghurt samples	pH	TA (%)	DM (%)	Ash (%)	Fat (%)	SNF (%)
Control	4.18±0.13 ^a	1.01±0.06 ^{abc}	18.72±0.37 ^a	0.68±0.01 ^a	4.76±0.08 ^a	13.96±0.37 ^a
А	4.14±0.13 ^a	0.92±0.05 ^{efgh}	16.48±0.43 ^{ghi}	0.54±0.04 ^g	4.08±0.14 ^{hi}	12.4±0.46 ^{efg}
В	4.16 ± 0.15^{a}	0.91±0.02 ^{fgh}	16.84±0.25 ^{fghi}	0.59 ± 0.02^{def}	4.24±0.20 ^{efgh}	12.6±0.36 ^{defg}
С	4.12 ± 0.08^{a}	$0.98{\pm}0.05^{abcde}$	17.9±0.63 ^b	0.62 ± 0.02^{bcd}	4.52±0.15 ^b	13.36±0.62 ^b
D	4.10 ± 0.12^{a}	0.90±0.03 ^{gh}	16.34±0.28 ^{hi}	0.57±0.03 ^{efg}	4.06±0.26 ⁱ	12.28±0.35 ^{efg}
E	4.10 ± 0.07^{a}	0.94±0.05 ^{defgh}	17.0±0.44 ^{efg}	0.58±0.03 ^{def}	4.28±0.15 ^{defg}	12.72±0.39 ^{cdef}
F	4.12 ± 0.08^{a}	0.95 ± 0.04^{cdefg}	17.52±0.55 ^{bcde}	0.62 ± 0.01^{bcd}	4.4 ± 0.07^{bcde}	13.12±0.61 ^{cbd}
G	4.16±0.15 ^a	0.92±0.03 ^{fgh}	16.3±0.39 ^{hi}	0.57±0.06 ^{efg}	4.12±0.08 ^{ghi}	12.18±0.40 ^{fg}
Н	$4.14{\pm}0.08^{a}$	$0.96{\pm}0.05^{bcdefg}$	17.1±0.37 ^{def}	$0.60{\pm}0.03^{def}$	4.36±0.11 ^{cdef}	12.74±0.39 ^{cdef}
Ι	$4.14{\pm}0.08^{a}$	1.0 ± 0.06^{abcd}	17.8±0.21 ^{bc}	0.62 ± 0.02^{bcd}	4.5±0.12 ^{bc}	13.3±0.14 ^{bc}
J	$4.14{\pm}0.08^{a}$	0.91 ± 0.04^{fgh}	16.28±0.94 ⁱ	0.56 ± 0.04^{fg}	4.2±0.07 ^{fghi}	12.08±0.88 ^g
Κ	4.12 ± 0.08^{a}	$0.96{\pm}0.06^{bcdef}$	17.2±0.29 ^{def}	0.60 ± 0.04^{cdef}	4.36±0.19 ^{cdef}	12.84±0.42 ^{bcde}
L	4.12 ± 0.08^{a}	0.95±0.05 ^{cdef}	17.98±0.26 ^b	0.64±0.03 ^{ab}	4.56±0.05 ^b	13.42±0.25 ^{ba}
М	4.2±0.12 ^a	$0.91 \pm 0.04^{\text{fgh}}$	16.86±0.33 ^{fgh}	0.57 ± 0.04^{efg}	4.14±0.13 ^{ghi}	12.72±0.30 ^{cdef}
Ν	4.14 ± 0.08^{a}	0.94±0.02 ^{defgh}	17.24±0.43 ^{cdef}	0.60 ± 0.02^{cde}	4.18±0.08 ^{ghi}	13.06±0.43 ^{bcd}
0	4.16±0.11 ^a	0.93±0.04 ^{efgh}	17.62 ± 0.40^{bcd}	0.63 ± 0.01^{bc}	4.44 ± 0.19^{bcd}	13.18±0.39 ^{bcd}
Р	4.2±0.12 ^a	$0.89{\pm}0.04^{h}$	16.48±0.36 ^{ghi}	0.59±0.01 ^{def}	4.08 ± 0.14^{hi}	12.4±0.45 ^{efg}
Q	4.18 ± 0.14^{a}	0.90±0.03 ^{gh}	17.1±0.39 ^{def}	0.61±0.03 ^{bcde}	4.26±0.15 ^{efg}	12.84±0.49 ^{bcde}
R	4.16±0.11 ^a	0.97±0.03 ^{bcdef}	17.58±0.75 ^{bcd}	0.65 ± 0.03^{ab}	4.36±0.11 ^{cdef}	13.22±0.75 ^{bc}

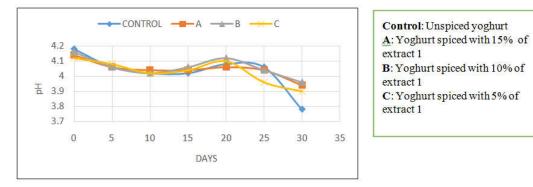
 $(^{a.b.c})$: The values with the same superscript letter in the same column are not significantly different (P>0.05).

TA: Titratable Acidity (%lactic acid); DM: Dry Matter; SNF: Non - Fat Solid

Yoghurtsamples	Colour	Odour	Taste	Texture	Overall acceptability
Control	4.00±0.81 ^{abcd}	3.61±1.04 ^{abcd}	3.82±0.90 ^{ab}	3.64±0.94 ^{abc}	3.76±0.95 ^{abcd}
Α	3.35±1.15 ^{efg}	3.55±1.07 ^{abcd}	3.00±1.07 ^{ghi}	$2.82{\pm}0.90^{h}$	3.23±0.88 ^{ef}
В	3.64 ± 0.88^{cdef}	3.29±0.93 ^{cde}	3.36±0.89 ^{cdefgh}	3.14±0.74 ^{efgh}	3.38 ± 0.69^{def}
С	3.91±0.75 ^{abcd}	3.55±1.10 ^{abcd}	3.64 ± 0.84^{abc}	3.35±0.88 ^{cdefg}	3.58±0.58 ^{bcde}
D	3.61±0.81 ^{cdef}	3.38±0.92 ^{bcde}	3.20±1.12 ^{defghi}	3.14±0.82 ^{efgh}	3.35±0.77 ^{ef}
Е	3.73±0.79 ^{cde}	3.55 ± 0.92^{abcd}	3.29±1.11 ^{cdefghi}	3.29±0.71 ^{defgh}	3.35±0.88 ^{ef}
F	3.76±0.92 ^{bcde}	3.67±1.03 ^{abc}	3.82±0.83 ^{ab}	3.61±0.92 ^{abcd}	3.88±0.84 ^{abc}
G	3.41±0.748 ^{efg}	3.20±1.06 ^{de}	$2.94{\pm}0.98^{hi}$	$3.05 \pm 0.88^{\text{fgh}}$	3.11 ± 0.87^{f}
Н	4.02±0.79 ^{abc}	3.76±0.92 ^{ab}	3.61±0,77 ^{abcd}	3.73±0.70 ^{abc}	3.85±0.78 ^{abc}
Ι	4.26±0.70 ^a	3.67±0.84 ^{abc}	$3,82\pm0.90^{ab}$	$3.94{\pm}0.85^{a}$	3.94±0.63 ^{ab}
J	2.85 ± 0.98^{hi}	3.00 ±0.69 ^e	3.08±0.75 ^{fghi}	2.97±0.86 ^{gh}	3.20±0.80 ^{ef}
K	3.11±0.97 ^{ghi}	3.23±0.85 ^{de}	3.20±0.80 ^{defghi}	3.11±0.84 ^{fgh}	3.26±0.82 ^{ef}
L	3.94 ± 0.77^{abcd}	3.52±0.78 ^{abcd}	3.58±0.85 ^{abcde}	3.61 ± 0.65^{abcd}	3.91±0.71 ^{abc}
М	2.79±0.94 ⁱ	3.23±0.95 ^{de}	2.91 ± 0.90^{i}	3.17±0.86 ^{efgh}	3.05±0.77 ^{fg}
Ν	3.23±0.98 ^{fgh}	3.38±0.85 ^{bcde}	3.38±0.88 ^{cdefg}	3.38±0.77 ^{cdef}	3.35±0.81 ^{ef}
0	3.58±0.89 ^{def}	3.32±0.87 ^{cde}	3.55±0.89 ^{abcde}	3.52±0.78 ^{bcde}	3.52±0.82 ^{cde}
Р	3.38±0.73 ^{efg}	3.32±0.68 ^{cde}	3.17±0.67 ^{efghi}	3.35±0.73 ^{cdefg}	3.20±0.6837 ^{ef}
Q	3.47±0.99 ^{efg}	3.32±1.06 ^{cde}	3.44±0.82 ^{bcdef}	3.41±0.74 ^{cdef}	3.55±0.92 ^{bcde}
R	4.17±0.71 ^{ab}	3.82±0.71 ^a	3.94±0.81 ^a	3.88 ± 0.68^{ab}	4.11±0.76 ^a

(a,b,c, ...): Les valeurs affectées des mêmes lettres sur une même colonne ne sont pas significativement différentes (P>0.05)

pH of ginger spiced yoghurt during 30 day of storage: Figures 1a to 1f show the changes of pH of yoghurt samples observed during 30 days of storage. The pH of the yoghurt samples both spiced and unspiceddid not indicate any significant (P>0.05) changes during storage except for the yoghurt spiced with 15% of ginger extract 3 (Figure 1c), yoghurt spiced with 5% of ginger extract 5 (Figure 1e) and yoghurt spiced with 15% of ginger extract 6 (Figure 1f) where during the last period (Days 25 - 30) of storage significant changes were observed.





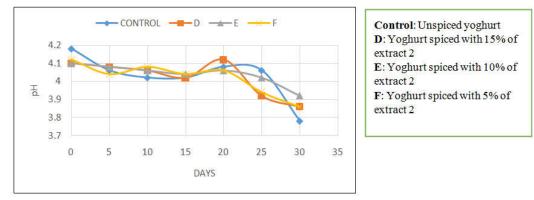
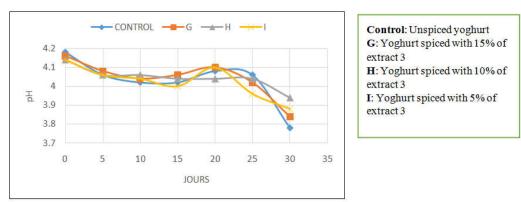


Figure 1b. Changes in pH levels of yoghurt spiced with extract 2 during 30days of storage





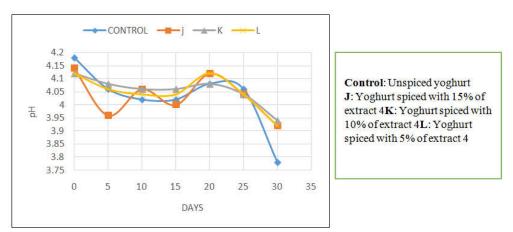
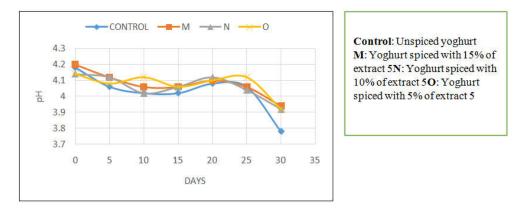
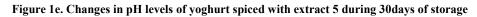
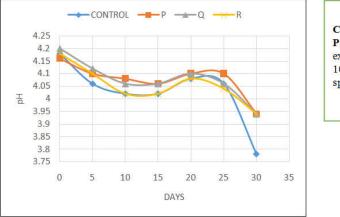
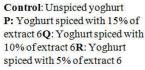


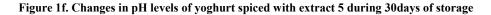
Figure 1d. Changes in pH levels of yoghurt spiced with extract 4 during 30days of storage

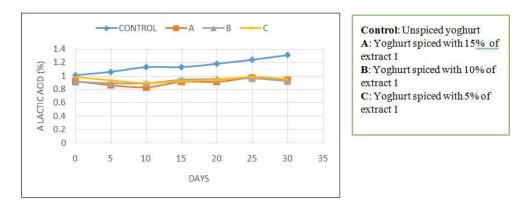














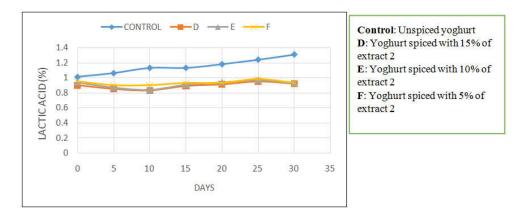
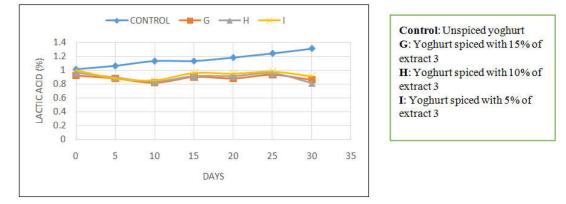
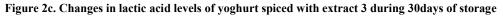
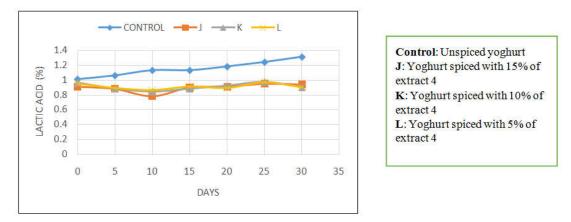


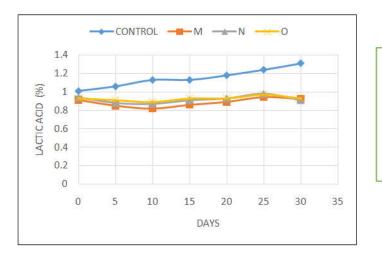
Figure 2b. Changes in lactic acid levels of yoghurt spiced with extract 2 during 30days of storage





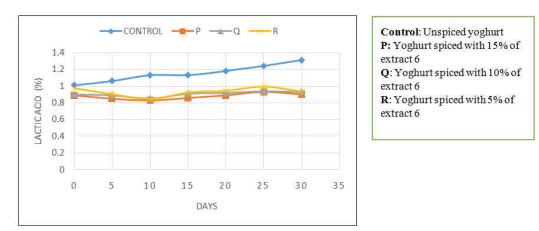






Control: Unspiced yoghurt **M**: Yoghurt spiced with 15% of extract 5**N**: Yoghurt spiced with 10% of extract 5**O**: Yoghurt spiced with 5% of extract 5





The pH value was unchanged till day 25 of storage and this may be related to the low activity of lactic acid bacteria. This insignificant change observed during pH evolution of all the voghurt samples during storage is in accordance with results of Duangrutai (2014) who used that fruits (pineapple, pawpaw and mango) to flavour yoghurt and stored for 21 days but not with that of Tarakçi and Küçüköner (2003), Salwaet al. (2004), Elfaki and AbbElrazig (2010), Isamet al. (2011), Manjulaet al. (2012), Yousef et al. (2013), Ayar and Gürlin (2014) and Njova et al. (2016) who observed reduction of yoghurt pH from the beginning of storage till 15 days of storage. Moreover, a significant decline (P<0.05) of pH between the initial and the final values of all the yoghurt samples was observed. The significant difference between the final and the initial values of pH of all the yoghurt samples during storage was expected and could be related to the continuous fermentation of lactose into lactic acid during storage at refrigerated conditions. During storage at refrigerated conditions, activity of lactic acid bacteria is not completely stopped and there is production of lactic acid leading to a drop of pH. At the end of storage, the unspiced yoghurt samples indicated the lowest (P<0.05) pH value while all the spiced yoghurt samples presented a comparable (P>0.05) pH values. This could indicate the low level of fermentation in spiced voghurt samples due to the antimicrobial activity of ginger extract. Ginger could thus contribute for the preservation of yoghurt.

Titratable acidity of ginger spiced voghurt during 30 day of storage: The changes in titratable acidity of the yoghurt samples are presented in Figures 2a - 2f. The unspiced yoghurt (control yoghurt sample) showed the highest (P<0.05) titratable acidity amongst all the yoghurt samples. Moreover, it indicated an increase of titratable acidity during storage except from day 10 to day 15 while all the other experimental yoghurt samples (spiced yoghurt samples) did not present significant (P>0.05) changes. Also, only the unspiced yoghurt sample indicated a significant difference (P≤0.05) between the final and the initial values during storage. At the end of storage, the highest (P≤0.05) titratable acidity was observed with the unspiced yoghurt sample and all the spiced yoghurt obtained the comparable (P>0.05) values. The increase of titratableacidity observed with the unspiced yoghurt sample was expected due to the action of lactic acid bacteria at storage conditions (4- 6°C) with production of lactic acid. This result is in conformity with previous findings (Tarakçi and Küçüköner, 2003; Salwa et al. 2004; Elfaki and AbbElrazig, 2010; Isam et al., 2011; Manjula et al., 2012; Yousef et al., 2013; Ayar and Gürlin, 2014 and Njoya et al., 2016). The unchanged values of titratable acidity observed during storage with spiced yoghurt samples could be explained by the inhibition of lactic acid bacteria activity (growth) due to the antimicrobial activity of ginger. The high acidity obtained with control yoghurt sample compare to experimental yoghurt samples at the end of storage could express a positive effect of ginger extract on the yoghurt preservation.

Sensory evaluation: The results of sensory evaluation presented in table 4 revealed that yoghurt sample I (Spiced yoghurt with 5% of extract 3) obtained the high score in term of colour. This value was not significant (P>0.05) different from those of control yoghurt (Unspiced yoghurt) sample and yoghurt samples C (Spiced yoghurt with 5% of extract 1), H (Spiced yoghurt with 10% of extract 3), L (Spiced yoghurt with 5% of extract 4) and R (Spiced yoghurt with 5% of

extract 6). Appreciation of the yoghurt colour with spiced yoghurt samples having 15% of ginger extract was low except sample D (Spiced yoghurt with 15% of extract 2). Thus, increasing the concentration of ginger extract in yoghurt lowered its appreciation by panellists. This could be due to the brownish coloration of the extract which modified the original colour of yoghurt. According to the panellists, all spiced voghurt samples showed similar (P>0.05) mean scores with the unspiced (control) yoghurt sample except for sample J (Spiced yoghurt with 15% of extract 4) which presented a lower mean score for odour. The taste of all the spiced yoghurt samples made with 5% of ginger extract was not significantly (P>0.05) different from each other and with that of the control though sample R (Spiced yoghurt with 5% of extract 6) obtained the high mean score. The increase in the concentration of extract in yoghurt led to reduction of the appreciation of the taste of yoghurt. Ginger is characterized by its pungent taste thus at high concentration, it could be highly expressed by reducing the taste of the spiced yoghurt. The texture of the control yoghurt sample was similar (P>0.05) to those of the spiced yoghurt with 5% of extract and different $(P \le 0.05)$ from those of the spiced yoghurt samples containing 15% of extract except sample P (Spiced yoghurt with 15% of extract 6). The quality of yoghurt texture decreased with increasing of the ginger extract concentration. The liquid form of the ginger could reduce the firmness of yoghurt and subsequently its texture by dilution. At lower proportion of extract, the dilution effect is low and could justify the similarity of texture between unspiced yoghurt and spiced voghurts with 5% of ginger extract. Yoghurt sample R (Spiced yoghurt with 5% of extract 6) presented mean score of overall acceptability which was not different (P>0.05) from those of yoghurt samples F, I and L respectively prepared with 5% of extract 2, 3 and 4. All the spiced yoghurt made with 5% of ginger extract showed the same level (P>0.05) of overall acceptability with the unspiced yoghurt sample which was not the case with yoghurt samples spiced with 15% of extract. The overall acceptability of yoghurt was generally reduced with extract concentration. increasing ginger At high concentrations, ginger sensory properties could be express strongly. Therefore, it affects the original sensory properties of yoghurt by lowering its appreciation.

Conclusion

Ginger extract incorporated to yoghurt have no effect on pH of yoghurt but decrease its titratable acidity, dry matter, fat, SNF and ash content and especially at high concentrations (10 and 15% (V/V)). During storage, they positively affect the change of pH and titratable acidity. Ginger extract, at 5% has no effect on the sensory properties of yoghurt and its overall acceptability and increasing its concentration lowers those properties.

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