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## IN VITRO DISSOLUTION STUDIES AND PHARMACEUTICAL EQUIVALENCE ASSESSMENT OF METFORMIN HYDROCHLORIDE TABLETS COMMERCIALY AVAILABLE IN SALVADOR, BAHIA, BRAZIL

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### ABSTRACT

Metformin is a drug widely used in worldwide for type 2 diabetes mellitus treatment. The aim of this study was to compare the dissolution profiles of Metformin hydrochloride tablets (850 mg) available as the reference drug (R), generic (G) and similar (S) commercialized in Bahia, Brazil using a simple, fast and low cost ultraviolet spectrophotometric method. The tests were performed in compliance with the Brazilian Pharmacopoeia and United States Pharmacopoeias. The proposed methodology for quantifying the drug was validated presenting precision, linearity and accuracy. All the products released metformin satisfactorily (>75% of the drug dissolved within 45 min). The product R, G and S showed dissolution efficiency of 83.85, 76.67 and 75.04%, respectively. From the calculation of similarity factor ( $f_2$ ), the formulations were considered pharmaceutical equivalents.

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### INTRODUCTION

Metformin is an oral anti-hyperglycemic of the biguanide class recommended as the first selection oral therapy for the treatment of type 2 diabetes mellitus (DM2), unless the patient has any renal impairment or other contraindications. This drug reduces insulin resistance and blood glucose levels by inhibiting hepatic gluconeogenesis, increasing the sensitivity of peripheral tissues to insulin. Clinical studies show its use in polycystic ovary syndrome, type 1 diabetes mellitus, as well as in obesity, including in adolescents with insulin resistance (ShurrabandArafa, 2020; Khomitskaya et al., 2018; Wang et al., 2017). Used as a hydrochloride, metformin is available in three manufacturing categories, in Brazil: reference, generic, and similar drugs. "Similar" drug products contains the same active ingredient and concentration, pharmaceutical form, route of administration, posology, and therapeutic indication. It is equivalent to the medicine registered in the federal agency responsible for sanitary surveillance, and may differ only in

characteristics related to the size and shape of the product, shelf-life, packaging, labeling, excipients, and vehicle. Similar drugs must always be identified by trade name or brand (Brasil, 2011). The aqueous solubility of metformin in the pH range of 1.2 to 6.8 is 300 mg/mL at 25 °C. However, 850 mg of metformin hydrochloride is dissolved more slowly in pH 1.2 and 4.5, than in pH 6.8 dissolution medium. It has two acid dissociation constants ( $pK_{a1} = 2.8$  and  $pK_{a2} = 11.5$ ) and, depending on pH, three different species are produced -  $\text{Metf}_1^{1+}$ ,  $\text{Metf}_2^{1+}$  and bioprotonated form  $\text{Metf}^{2+}$  (Patiño-herrera et al., 2019). It is hypothesized that the additional protonation of metformin at acidic pH results in greater solvation and a larger hydrodynamic radius, leading to slower diffusion and dissolution. Belongs to class 3 of the biopharmaceutical classification system (BCS), with high water solubility and low permeability to cell membranes (Wang et al., 2017; Desai et al., 2014; Patiño-herrera et al., 2019). The main site of absorption of metformin is the proximal part of small intestine. It has absolute oral bioavailability (50 to 60%),

peak plasma concentration (approximately 3 h) and plasma elimination half-life (2.0 to 6.0 h). The biological half-life of metformin hydrochloride is 1.5 to 1.6 hours and, the drug is commonly administered in high doses (500 or 1000 mg), in 2 to 3 times a day, to obtain an effective treatment efficacy (Kim and Park, 2015; Oh *et al.*, 2013). Oral route is the most commonly used for administration of solid pharmaceutical forms. Factors, such as particle size, excipients, compaction strength, dosage form, drug release profile, gastric emptying, gastrointestinal transit time, and drug absorption location can affect the drug release rate. Drug dissolution and release are critical aspects of quality, which can affect bioavailability and pharmacokinetics, which implies the efficacy and safety of drugs. Therefore, the dissolution test is used to guarantee the *in vitro* similarity of dissolution between two drugs, which is related to pharmaceutical equivalence and, consequently, to bioavailability and bioequivalence between products (Oh *et al.*, 2013; Dumarey *et al.*, 2015; Ariyasu *et al.*, 2016; Pawar *et al.*, 2016; Dias *et al.*, 2020). Regulatory agencies and several pharmacopoeias, such as Brazilian (BP, 2019) and United States (USP, 2018), recommend the use of UV spectrophotometry for the quantitative analysis of the drugs in dissolution test samples, ensuring that a specified amount of the active pharmaceutical ingredient is dissolved at a predefined time. In the present study, were obtained *in vitro* dissolution profiles using a UV spectrophotometric method to evaluate the release of tablets containing 850 mg of metformin (reference, generic and similar drugs), commercially available in Salvador, Bahia, Brazil. Dissolution profiles were compared by similarity factors ( $f_2$ ) and the area under a dissolution curve (AUC) to assess pharmaceutical equivalence.

## MATERIAL AND METHODS

**Material (chemicals/reagents and samples):** All reagents were of analytical grade. Monobasic potassium phosphate ( $\text{KH}_2\text{PO}_4$ ) and, dibasic potassium phosphate ( $\text{K}_2\text{HPO}_4$ ) were obtained from Neon Comercial (São Paulo, SP, Brazil) and used as the dissolution medium for the dissolution tests. Doubly distilled water was obtained from a Q341 Quimis distiller (São Paulo, SP, Brazil) and used throughout the experiments. All laboratory material was washed in a 10% (v/v)  $\text{HNO}_3$  solution bath for 24 h, washed with high purity water and dried at room temperature. Metformin reference standard was obtained from National Institute for Quality Control in Health (INCQS), at Oswaldo Cruz Foundation (Fiocruz), indicated by the Brazilian Pharmacopoeia.

Metformin tablets (reference, generic, and similar products) containing 850 mg were purchased from three drug stores in Salvador, Bahia, Brazil. The metformin tablets were subjected to weight variation, friability, disintegration, and dissolution tests in accordance with Brazilian and United States Pharmacopoeias general methods applied to drugs. All tests were performed on products within the expiration date. The products (labeled to contain 850 mg of the drug) were designated as: R (reference product) and the following excipients: hypromellose, magnesium stearate and, povidone.; G (generic product) and the following excipients: hypromellose, magnesium stearate, povidone, polyethylene glycol, ethyl alcohol, croscarmellose sodium, microcrystalline cellulose and, silicon dioxide; and, S (similar product) and the following excipients: hypromellose, magnesium stearate, povidone, polyethylene glycol, polysorbate 80 and, purified water.

**Calibration standards:** A reference stock solution of 1000  $\mu\text{g/mL}$  was prepared with the reference chemical. Calibration standards with concentrations ranging from 50 to 1000  $\mu\text{g/mL}$  were prepared daily from the standard solution by appropriate dilution, stored in decontaminated polyethylene bottles and analyzed in triplicate by UV spectrophotometry at  $\lambda = 233 \text{ nm}$ .

**Physical Tests (Weight Uniformity, Friability and Disintegration):** For determination of weight uniformity, an analytical balance M164-AI (Mark, Piracicaba, SP, Brazil) was used. For each product (reference, generic and similar) containing 850-mg of metformin, 20 tablets were selected at random and weighed individually on the analytical balance. Then, the average weight, standard deviation and individual deviation of each tablet from the mean for the three products (reference, generic and similar) were determined. The Brazilian Pharmacopoeia (2019) and United States Pharmacopoeia (USP) establishes a maximum variation of  $\pm 5.0\%$ . No more than two units outside the specified limits can be tolerated in relation to the average weight. However, none can be above or below twice the indicated percentages. Friability was determined on friabilometer HX 300-2 (Ethik, Vargem Grande Paulista, SP, Brazil). For the friability test, 10 metformin tablets (850 mg) were weighed and submitted to the friabilometer at a speed of 25 rpm for 4 minutes (100 cycles). After this period, the powders were removed and the tablets were weighed again. With the difference in weights, the percentage of loss was calculated and, compared with the value established in the Brazilian and USP pharmacopoeias ( $\leq 1.5\%$  of weight loss) (BF, 2019). Disintegrator 301/AC 01 system (Nova Ética, Vargem Grande Paulista, SP, Brazil) and a multiple bath dissolver ( $n = 6$ ) (Ethik, 299, Vargem Grande Paulista, SP, Brazil) were used for tests. In the disintegration test apparatus, six units of each sample (tablets) of each brand were used under the following conditions: distilled water as disintegration medium at  $37 \text{ }^\circ\text{C} \pm 1 \text{ }^\circ\text{C}$  and time of 30 minutes, in accordance with Brazilian Pharmacopoeia (2019) and United States Pharmacopoeia (USP).

**Invitro Physicochemical Equivalence (Dissolution Studies):** A PG 2000pH meter (Gehaka, , São Paulo, SP, Brazil) was used to determine the pH of dissolution medium and all spectrophotometric measurements were performed using a UV-Visible spectrophotometer (Shimadzu, 1240 model, Kyoto, Japan) equipped with a diode matrix detector (DAD), which was set at  $\lambda = 233 \text{ nm}$  and, the absorbances of sample solutions were read in 1-cm quartz cells, in triplicate. Dissolution tests were performed in the dissolver Model 299 (Ethik, Vargem Grande Paulista, SP, Brazil) with six replicates, in 900 mL of the dissolution medium, phosphate buffer (pH 6.8) at a temperature of  $37 \pm 0.5 \text{ }^\circ\text{C}$  and rotation of 100 rpm with apparatus 1 (basket), as described in Brazilian pharmacopoeia. To obtain dissolution profiles, 10 mL sample aliquots were taken at predetermined time intervals (1, 3, 5, 10, 15, 20, 30, 45 and 60 min) and replaced with an equal volume of the dissolution medium for maintain a constant total volume. The collected aliquots were filtered (0.44  $\mu\text{m}$  membrane filters) and transferred to amber flasks for later reading by UV spectrophotometry at  $\lambda = 233 \text{ nm}$ . The absorbances were converted into concentrations obtained from the equation on the standard curve. The calculations were performed considering the amount of drug removed in each aliquot. The results were expressed as a percentage as a function of time. Dissolution profiles were compared to two methods: similarity

factor ( $f_2$ ) and dissolution efficiency (ED), in accordance to RDC 31/2010, which provides for Pharmaceutical Equivalence Studies and Comparative Dissolution Profile. The comparison of dissolution profiles is useful in cases where it is desired to know the behavior of two drugs before submitting them to a study of relative bioavailability / bioequivalence. Comparative dissolution profiles are evaluated only using the calculation of the similarity factor ( $f_2$ ), which corresponds to a measure of similarity between the dissolved percentages of both profiles (Brasil, 2010). The calculation of the similarity factor ( $f_2$ ) was performed using the equation:

$$f_2 = 50 \times \log \left\{ \left[ 1 + \left( \frac{1}{n} \right) \sum_{i=1}^n (R_t - T_t)^2 \right]^{-0.5} \times 100 \right\}$$

$n$  = number of data points,  $R_t$  and  $T_t$  = % dissolved, from each of the two formulations at a specific time point. The dissolution profiles are considered similar if  $f_2$  is  $\geq 50$  (Al durdunji *et al.*, 2016). From dissolution profiles, the dissolution efficiency (DE) was calculated by the percentage ratio of the area under the dissolution curve up to time  $t$  in relation to the area of the rectangle described by 100% dissolution at the same time (Simionato *et al.*, 2018). The calculation of DE was performed using the equation:

$$\%DE = \frac{AUC_0^t}{Q_{100,t}} \times 100$$

DE values were analyzed using the Student's t test with a significance level of  $p \leq 0.05$  and, DE of tested products were subjected to analysis of variance (ANOVA), using the Statistical Package for the Social Sciences (SPSS) software (version 22.0, IBM).

**Analytical validation:** The method for dissolution test was validated following the Brazilian Resolution 166/2017 (Brazil, 2017) and International Harmonization Conference - ICH (ICH, 2005). Selectivity, linearity, precision (repeatability and intermediate precision), accuracy, limit of detection (LOD) and quantification (LOQ) were determined. Possible interferences were assessed by testing the components of products (excipients) of the formulations and comparing them with a placebo. Linearity was obtained through the correlation coefficient, by linear regression analysis using the least squares method, of three analytical curves with seven different concentrations, ranging from 100 to 1000  $\mu\text{g/mL}$  of metformin. Repeatability and intermediate precision were performed to assess the precision of method. The relative standard deviation (RSD) was used to calculate the quantification precision of three known concentrations of metformin (50, 400 and 900  $\mu\text{g/mL}$ ) on the same and alternate days. Accuracy was expressed as the agreement between the defined reference value and the measured value of each concentration, in triplicate, by the UV-VIS spectrophotometer.

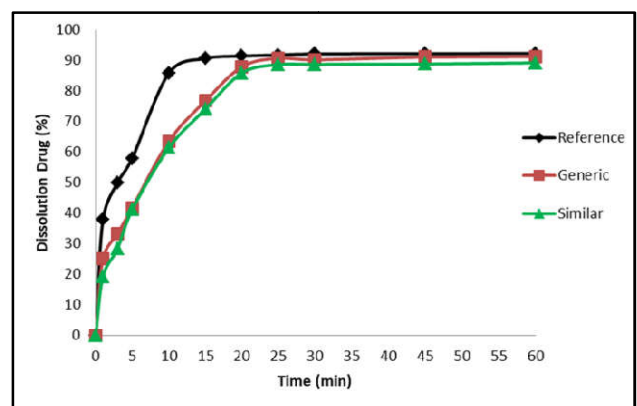
## RESULTS AND DISCUSSION

**Analytical method validation:** The method was validated to demonstrate selectivity, linearity, LOD, LOQ, precision and accuracy (Table 1). There was no interference from the matrix components (excipients). The standard calibration curve resulting from the working concentration range (50.0 to 1000.0  $\mu\text{g/mL}$ ) was obtained with a correlation coefficient ( $r$ ) above (0.99). UV spectrophotometry method proposed in this study

confirmed the ability of rapid analysis with good repeatability of this analytical technique for the determination of metformin in pharmaceutical products. [Table 1. Evaluation of analytical parameters for metformin determinations, in tablets.]

**Table 1. Evaluation of analytical parameters for metformin determinations, in tablets**

Analytical parameter	Results
Regression equation	$y = 0.0963x - 0.0047$
Linear range ( $\mu\text{g/mL}$ )	50 – 1000
Correlation coefficient ( $r$ )	0.9981
Limit of detection (LOD), in $\mu\text{g/mL}$	0.8
Limit of quantification (LOQ), in $\mu\text{g/mL}$	1.9
Precision* (%)	3.57 – 5.78
Accuracy* (%)	97.82 – 100.17



**Figure 1. Comparative dissolution profiles of metformin (850 mg) tablets of reference, generic, and similar products. (USP type 1 apparatus at 100 rpm with 900 mL phosphate buffer (pH 6.8) at  $37.0 \pm 0.5$  °C for 1 h.)**

**Physical Tests (Weight Uniformity, Friability and Disintegration):** The weight variation test is applicable to solid oral dosage forms (capsules, uncoated and film-coated tablets), and makes it possible to check whether the units in the same batch have uniform weight. The results showed that the R, G and S products had satisfactory average weight values (0.8925; 1.0053 and 0.9509 mg, respectively), according to the Brazilian and USP pharmacopoeias (variation of  $\pm 5.0\%$ ). For generic drug, there was a greater variation in weight, which can be explained by the irregular diameter of the tablets. Friability is the tendency for a tablet to lose component particles due to abrasion, friction or mechanical shock. High friability leads to unacceptable loss of drug content during storage and handling. In addition to the potential loss of therapeutic effects due to underpower, the damaged appearance of the tablet also raises doubts about the quality of the tablet. The test applies only to uncoated tablets (Osei-yeboahand Sun, 2015;BF, 2019). R, G and S products showed friability within the limit established by the Brazilian pharmacopeia (BF, 2019), which indicates a loss of mass equal to or less than 1.5% of its weight. Disintegration is a physical phenomenon, where tablets break down into smaller particles after exposure to fluids, leading to an increase in the surface area of the dosage form. It is promoted by the presence of disintegrants in the formulation, allowing greater wettability and expansion of the tablet size. The pressure generated by the expansion of the size creates micro-cracks that propagate, favoring the entry of additional water and, finally, leading to complete rupture (Bisharat *et al.*, 2019). All products tested met the Brazilian and USP pharmacopoeias requirements for tablets, i.e., disintegration time less than 30 minutes.

Reference, generic and similar products disintegrated in 8'13"; 9'35" and 9'26", respectively. The results of this study showed disintegration times longer than those found by Venkateswara Rao *et al.* (2017), in range of 4'58" to 8'45", in metformin hydrochloride tablets in Vijayawada, Índia.

**Dissolution Test, Dissolution Profiles (DP) and Dissolution Efficiency (DE):** Brazilian and USP pharmacopoeias recommends that at least 75% of the amount of metformin declared on the product label (850 mg) be dissolved in 45 minutes. *In vitro* dissolution studies are important for the development, characterization of pharmaceutical forms and prediction of the respective *in vivo* performance (Capková-helesicová *et al.*, 2019). Multipoint analysis graphs (Figure 1) were plotted to assess the amount of the drug dissolved versus the time of the metformin tablets (reference, generic and similar). [Figure 1. Comparative dissolution profiles of metformin (850 mg) tablets of reference, generic, and similar products. (USP type 1 apparatus at 100 rpm with 900 mL phosphate buffer (pH 6.8) at 37.0 ± 0.5 °C for 1 h.)]. Products R, G and S showed comparable levels of metformin (850mg), in 45 minutes, have dissolved and released the drug in accordance with the Brazilian and USP pharmacopoeias (>75% of drug within 45 min), with 92.3, 91.2 and 88.9%, respectively. Across dissolution profiles obtained was noted that the drugs (similar and generic) released metformin more slowly at the onset of process, when compared with the reference drug. It is suggested that this difference may be associated with the presence of polyethylene glycol in these products. In solid dosage formulations, polyethylene glycols, due to their higher molecular weight, can increase the effectiveness of tablet binders and give plasticity to the granules, which can prolong disintegration if present in concentrations above 5% w/w (Rowe *et al.*, 2009). The results obtained for similar product, in this study, were contradictory with Sevenine and (2016), which compared dissolution profiles of drugs (reference and similar) containing metformin (850 mg), marketed in Brazil. The authors found amounts of 95.66% and 20.32% of metformin released, in 45 minutes, respectively, for the reference and similar products. These results indicate poor quality of similar tablets, in disagreement with the Brazilian and USP pharmacopoeias (Silva *et al.*, 2016). On the other hand, the data were in agreement with Venkateswara Rao *et al.* (2017), in their studies conducted in India. These authors obtained release (> 70%) of metformin hydrochloride from all brands of tablets, within 45 minutes.

Dissolution profiles can be analyzed by mathematical methods, using the independent method using the parameters: AUC and  $f_2$  (similarity factor). The application of these different methods to compare dissolution profiles can provide more accurate information on the dissolution behavior of products (Viana *et al.*, 2015). The  $f_2$  calculation was performed for generic and similar products in relation to the reference drug, as they did not present a very fast dissolution (they did not reach more than 85% in the period of up to 15 minutes). Both products (similar and generic) showed similarity over 50 and, therefore, were considered pharmaceutical equivalents (Oliveira *et al.*, 2014). Dissolution efficiency (DE%) is a parameter widely used to assess pharmaceutical equivalence between formulations. Through DE, it is allowed to evaluate not only the amount of drug released after a certain time, but the release kinetics throughout the period in question. This concept has the advantage of being theoretically related to *in vivo* data, since the extent of absorption of a drug *in vivo* is proportional to its

dissolved concentration and the time it remains in contact with the absorption regions of the gastrointestinal tract (Oliveira *et al.*, 2014). From dissolution profiles of each product, it was possible to calculate DE (83.85%, 76.67% and 75.04%, for R, G and S, respectively).

## Conclusion

Metformin is a drug widely used in Brazil and worldwide for type 2 diabetes mellitus treatment. Products R, G and S showed comparable levels of metformin (850 mg), in 45 minutes, have dissolved and released the drug in accordance with the Brazilian and USP pharmacopoeias (>75% of drug within 45 min). From the calculation of the similarity factor ( $f_2$ ), the formulations were considered pharmaceutical equivalents. In the comparative studies of the dissolution profile, the reference drug released more than 87% drug reached in 15 min. Differences observed in the disintegration times and in the dissolution profiles for similar and generic products may have been influenced by the drug production process and the presence of excipients that delay dissolution in these specialties. For dissolution studies for metformin tablets a rapid and sensitive UV spectrophotometric method was developed with good linearity, precision, and accuracy.

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## Conflict Of Interest

The authors declare that there are not conflicts of interest regarding this manuscript.

## REFERENCES

- Al durdunji, A., Alkhatib, H. S., Al-ghazawi, M. 2016. Development of a biphasic dissolution test for Deferasirox dispersible tablets and its application in establishing an *in vitro*-*in vivo* correlation. *European Journal of Pharmaceutics and Biopharmaceutics*, 102, pp 9-18
- Ariyasu, A., Hattori, Y., Otsuka, M. 2016. Delay effect of magnesium stearate on tablet dissolution in acidic medium. *International Journal of Pharmaceutics*, 511, pp 757-764
- Bisharat, L., Alkhatib, H. S., Muhaisen, S., Quodbach, J., Blaibleh, A., Cespi, M., Berardi, A. 2019. The influence of ethanol on superdisintegrants and on tablets disintegration. *European Journal of Pharmaceutical Sciences*, 129, pp 140-147
- BP. Brazilian Pharmacopeia, 6th ed.; Brasília: Brazilian Health Surveillance Agency; 2019.
- Brasil. Ministry of Health, Brazilian Health Surveillance Agency ANVISA 2017. Resolution of the Board of Directors – RDC No. 166.: Brasília, July 24, 2017. Accessed Feb 20, 2020. <https://www20.anvisa.gov.br/coifa/pdf/rdc166.pdf>.
- Brasil. Ministry of Health, Brazilian Health Surveillance Agency ANVISA 2017. Concepts and definitions of medicines.: Brasília, 2011. Accessed Mar 28, 2020. <http://portal.anvisa.gov.br/medicamentos/conceitos-e-definicoes>

- Brasil. Ministry of Health, Brazilian Health Surveillance Agency ANVISA 2010. Resolution of the Board of Directors – RDC No. 31.: Brasília, August 11, 2010. Accessed Feb 23, 2020. [http://portal.anvisa.gov.br/documents/33880/2568070/res0031\\_11\\_08\\_2010.pdf/5e157d15-d3d5-4bb9-98db-5667e4d9e0c8](http://portal.anvisa.gov.br/documents/33880/2568070/res0031_11_08_2010.pdf/5e157d15-d3d5-4bb9-98db-5667e4d9e0c8).
- Capková-helesicová, T., Pekárek, T., Schöngut, M., Matejka, P. 2019 New designed special cells for Raman mapping of the disintegration process of pharmaceutical tablets. *Journal of Pharmaceutical and Biomedical Analysis*, 168, pp 113-123
- Desai, D., Wong, B., Huang, Y., Ye, Q., Tang, D., Guo, H., Huang, M., Timmins, P. 2014. Surfactant-Mediated Dissolution of Metformin Hydrochloride Tablets: Wetting Effects Versus Ion Pairs Diffusivity. *Journal of Pharmaceutical Sciences*, 103, pp 920-926
- Dias, F.S., Teles Júnior, Gilmar A.C., Oliveira, J.L.S., Bonfim, D.A., Santos, J.A., Souza, L.B.S., Oliveira, A.S., Dias, F.S., Santos Júnior, A.F. 2020. Development of new methodology for in vitro dissolution evaluation of marketed amlodipine besylate tablets in salvador/bahia/brazil employing factorial experimental design. *Brazilian Journal of Development*, 6, pp 14684-14703
- Dumarey, M., Goodwin, D.J., Davison, C. 2015. Multivariate modelling to study the effect of the manufacturing process on the complete tablet dissolution profile. *International Journal of Pharmaceutics*, 486, pp 112-120
- ICH. International Conference on Harmonisation ICH of Technical Requirements for Registration of Pharmaceuticals for Human Use. *Validation of Analytical Procedures: Text and Methodology Q2 R1*; ICH Harmonised Tripartite Guideline: Geneva, Switzerland, 2005. Accessed Feb 23, 2020. [https://database.ich.org/sites/default/files/Q2\\_R1\\_Guideline.pdf](https://database.ich.org/sites/default/files/Q2_R1_Guideline.pdf).
- Khomitskaya, Y., Tikhonova, N., Gudkov, K., Erofeeva, S., Holmes, V., Dayton, B., Davies, N., Boulton, D. W. and Tang, W. 2018. Bioequivalence of Dapagliflozin/Metformin Extended-release Fixed-combination Drug Product and Single-component Dapagliflozin and Metformin Extended-release Tablets in Healthy Russian Subjects. *Clinical Therapeutics*, 40, pp 550 - 561
- Kim, D.W. and Park, J.B. 2015. Development and pharmaceutical approach for sustained-released metformin succinate tablets. *Journal of Drug Delivery Science and Technology*, 30, pp 90 - 99
- Oh, T.O., Kim, J.Y., Ha, J.M., Chi, S.C., Rhee, Y.S., Park, C.W., Park, E.S. 2013. Preparation of highly porous gastroretentive metformin tablets using a sublimation method. *European Journal of Pharmaceutics and Biopharmaceutics*, 83, pp 460-467
- Oliveira, P.R., Bernardi, L.S., Silva, I.R.; Plácido, C.G.; Cardoso, S.G.; Silva, M.A.S.; Murakami, F.S. 2014 Avaliação da qualidade e equivalência farmacêutica de comprimidos contendo 10mg de sinvastatina. *Journal of Basic and Applied Pharmaceutical Sciences*, 35, pp 393-400
- Osei-yeboah, F. and Sun, C.C. 2015. Validation and applications of an expedited tablet friability method. *International Journal of Pharmaceutics*, 484, pp 146-155
- Patiño-herrera, R., Louvier-hernández, J.F., Escamilla-silva, E.M., Chaumel, J., Escobedo, A.G.P., Pérez, E. 2019 Prolonged release of metformin by SiO<sub>2</sub> nanoparticles pellets for type II diabetes control. *European Journal of Pharmaceutical Sciences*, 131, pp 1-8
- Pawar, P., Wang, Y., Keyvan, G., Callegari, G., Cuitino, A., Muzzio, F. 2016. Enabling real time release testing by NIR prediction of dissolution of tablets made by continuous direct compression CDC. *International Journal of Pharmaceutics*, 512, pp 96-107
- Rowe, R.C.; Sheskey, P.J.; Quinn, M.E. 2009. *Handbook of pharmaceutical excipients*, 6rd ed; London: *Pharmaceutical Press*, pp 1-917.
- Shurrah N.T. and Arafá E.S.A. 2020 Metformin: A review of its therapeutic efficacy and adverse effects. *Obesity Medicine*. 17, pp 180-186
- Silva, S., Roney, B.R. 2016. Estudio del perfil de disolución de metformina de referencia y de un similar. *Revista Cubana de Farmacia*, 50, pp 1561-2988
- Simionato, L.D., Petrone, L., Baldut, M., Bonafede, S.L., Segall, A.I. 2018 Comparison between the dissolution profiles of nine meloxicam tablet brands commercially available in Buenos Aires, Argentina. *Saudi Pharmaceutical Journal*, 26, pp 578-584
- USP. The United States Pharmacopeia and National Formulary USP 41-NF 3640; The United States Pharmacopeial Convention, Inc.: Rockville, MD, 2018.
- Venkateswara Rao, S., Divya, K., Padmalatha, K. 2017. Comparative in vitro studies and bioequivalence assessment of some commercially available metformin hydrochloride tablets in Vijayawada. *International Journal of Development Research*, 7, pp 16534-16540
- Viana, L.C.M.G., Ferreira, M.S., Mota, M.D., Magalhães, H.I.F., Júnior, A.F.S. 2015. Study of Dissolution Profiles and Disintegration of Tablets Containing Hydrochlorothiazide Marketed in Bahia, Brazil. *Latin American Journal of Pharmacy*, 34, pp 2010-2015
- Wang, C., Hu, S., Sun, C.C. 2017. Expedited development of a high dose orally disintegrating metformin tablet enabled by sweet salt formation with acesulfame. *International Journal of Pharmaceutics*, 532, pp 435-443

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